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## **ABSTRACT**:

## Silica-based porous catalysts for efficient CO<sub>2</sub> capture and methanation: a closer look at the surface.

L. Castoldi<sup>1</sup>, B. Di Credico<sup>2</sup>, R. Scotti<sup>2</sup>, L. Lietti<sup>1</sup>, C. Cristiani<sup>3</sup>, <u>E. Finocchio</u><sup>4</sup>.

<sup>1</sup>Dept. of Energy, Politecnico di Milano, 20156 Milano, IT.

<sup>2</sup>Dept. of Materials Science, Università degli Studi di Milano-Bicocca, 20126 Milano, IT. <sup>3</sup>Dept. of Chemistry, Material and Chemical Engineering "G. Natta", Politecnico di Milano, 20133 Milano, IT.

<sup>4</sup>Dept. of Civil, Chemical and Environmental Engineering, Università di Genova, 16145 Genova, IT.

Dual function materials are becoming more and more appealing to develop efficient and costeffective CO<sub>2</sub> capture, storage and re-use catalytic systems in a decarbonization approach. In this field, mesoporous silicas can be proposed as an alternative to the widely investigated alumina catalytic support [1]. This presentation will focus on a series of Ru/Ba/SiO<sub>2</sub> ternary catalysts, developed in form of powders synthetized by sequential impregnation of commercial silica, and tested in CO<sub>2</sub> adsorption and methanation process. IR spectroscopic investigations highlighted the complex properties of these surfaces, characterized by basic sites, weakly acidic silanol groups and the formation of a complex population of carbonated species following CO<sub>2</sub> adsorption-desorption experiments. In the framework of the SILCO project [2], the optimized formulations selected are now being applied to the preparation of catalysts based on silica from wastes, namely silica derived from hexafluorosilicic acid, a hazardous by-product of the phosphate and aluminum fluoride industries. Through a precipitation step, and the incorporation of organic templates, mesoporous and macroporous silica were produced, with diversified morphology and surface properties consistent with commercial materials. Silica from wastes can be promising candidates as support for methanation catalysts.

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A. Porta, R. Matarrese, C. Visconti and L. Lietti, Energy Fuels, 37, 7280 (2023).
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